Advisory Action Before the Filing of an Appeal Brief

Application No.	Applicant(s)	
10/584,725	KITAOKA ET AL.	
Examiner	Art Unit	
MATTHEW J. SONG	1792	

	MATTHEW J. SONG	1792		
The MAILING DATE of this communication appe	ars on the cover sheet with the o	correspondence add	ress	
THE REPLY FILED 28 January 2009 FAILS TO PLACE THIS APPLICATION IN CONDITION FOR ALLOWANCE.				
1. The reply was filed after a final rejection, but prior to or on application, applicant must timely file one of the following application in condition for allowance; (2) a Notice of Apperfor Continued Examination (RCE) in compliance with 37 C periods:	the same day as filing a Notice of a replies: (1) an amendment, affidavi eal (with appeal fee) in compliance	Appeal. To avoid abar t, or other evidence, w with 37 CFR 41.31; or	hich places the (3) a Request	
a) The period for reply expires <u>3</u> months from the mailing date	of the final rejection.			
b) The period for reply expires on: (1) the mailing date of this A no event, however, will the statutory period for reply expire la Examiner Note: If box 1 is checked, check either box (a) or (MONTHS OF THE FINAL REJECTION. See MPEP 706.07(1)	ater than SIX MONTHS from the mailing b). ONLY CHECK BOX (b) WHEN THE	g date of the final rejectio	n.	
Extensions of time may be obtained under 37 CFR 1.136(a). The date have been filed is the date for purposes of determining the period of extunder 37 CFR 1.17(a) is calculated from: (1) the expiration date of the set forth in (b) above, if checked. Any reply received by the Office later may reduce any earned patent term adjustment. See 37 CFR 1.704(b). NOTICE OF APPEAL	on which the petition under 37 CFR 1.1 ension and the corresponding amount on hortened statutory period for reply origing than three months after the mailing dat	of the fee. The appropria nally set in the final Offic	te extension fee e action; or (2) as	
 The Notice of Appeal was filed on A brief in comp filing the Notice of Appeal (37 CFR 41.37(a)), or any exter Notice of Appeal has been filed, any reply must be filed with AMENDMENTS 	nsion thereof (37 CFR 41.37(e)), to	avoid dismissal of the		
	out prior to the date of filing a brief	will not be entered be	called	
 The proposed amendment(s) filed after a final rejection, but prior to the date of filing a brief, will <u>not</u> be entered because (a) They raise new issues that would require further consideration and/or search (see NOTE below); 				
(b) They raise the issue of new matter (see NOTE below		•		
(c) They are not deemed to place the application in bet	ter form for appeal by materially red	ducing or simplifying th	ne issues for	
appeal; and/or (d) ☐ They present additional claims without canceling a c	corresponding number of finally reje	ected claims		
NOTE: (See 37 CFR 1.116 and 41.33(a)).	serresperialing frameer or initially reju	ottod oldiirio.		
4. The amendments are not in compliance with 37 CFR 1.12	21. See attached Notice of Non-Co	mpliant Amendment (F	PTOL-324).	
5. Applicant's reply has overcome the following rejection(s):			•	
 Newly proposed or amended claim(s) would be all non-allowable claim(s). 	owable if submitted in a separate, t	timely filed amendmer	t canceling the	
7. For purposes of appeal, the proposed amendment(s): a) [how the new or amended claims would be rejected is prov The status of the claim(s) is (or will be) as follows:		I be entered and an ex	planation of	
Claim(s) allowed: Claim(s) objected to:				
Claim(s) objected to: Claim(s) rejected:				
Claim(s) withdrawn from consideration:				
AFFIDAVIT OR OTHER EVIDENCE				
 The affidavit or other evidence filed after a final action, bubecause applicant failed to provide a showing of good and was not earlier presented. See 37 CFR 1.116(e). 				
 The affidavit or other evidence filed after the date of filing entered because the affidavit or other evidence failed to o showing a good and sufficient reasons why it is necessary 	vercome <u>all</u> rejections under appea	al and/or appellant fails	to provide a	
10. ☐ The affidavit or other evidence is entered. An explanation REQUEST FOR RECONSIDERATION/OTHER	n of the status of the claims after er	ntry is below or attache	ed.	
 The request for reconsideration has been considered but See Continuation Sheet. 	t does NOT place the application in	condition for allowand	ce because:	
12. ☐ Note the attached Information <i>Disclosure Statement</i> (s). (13. ☐ Other:	PTO/SB/08) Paper No(s)			
	/Robert M Kunemund/ Primary Examiner, Art U	Init 1792		

Continuation of 11. does NOT place the application in condition for allowance because: Applicant's arguments filed 1/28/2009 have been fully considered but they are not persuasive.

In response to applicant's argument that the addition of Mg improves reproducibility of the crystal growth, the fact that applicant has recognized another advantage which would flow naturally from following the suggestion of the prior art cannot be the basis for patentability when the differences would otherwise be obvious. See Ex parte Obiaya, 227 USPQ 58, 60 (Bd. Pat. App. & Inter. 1985). Applicant teaches on page 3 of the specification that the flux includes Mg and the crystal can have a p-type electrical characteristic and the amount of nitrogen dissolved in the melt is increased because the flux includes Mg (paragraph [0012]). Therefore, Applicant teaches the Mg has two functions, one to dope and one as a flux to increase amount of dissolved nitrogen. Jeong et al teaches the basic mixture is one among magnesium doped to impurity (ET paragraph [26]). Jeong et al is silent to the concentration of the magnesium and Yoshida et al teaches doping with Mg to form a p-type GaN having a desired electrical resistivity, where that amount of dopant overlaps the range taught by applicant. Therefore, the proportion of Mg to the sum of Mg and Na would naturally flow from the teachings of the prior art because the prior art teaches overlapping impurities concentrations in the grow crystal thus would require optimization of the amount of Mg and the amount of Mg is a dopant, thus would tend to be a small proportion of the melt mixture.

It is noted that differences in concentration will not support the patentability of subject matter encompassed by the prior art unless there is evidence indicating such concentration or temperature is critical (MPEP 2144.05). The combination of Jeong et al and Yoshida et al teaches doping Mg into the GaN crystal to obtain a doped crystal, and the amount of dopant within the mixture can be determined through routine experimentation to obtain the desired dopant amount. The combination of Jeong et al and Yoshida et al teaches a similar process of crystal growth, as applicant, comprising Na and Mg, where Mg is supplied as a doping impurity and the amount of Mg in the crystal overlaps; therefore the amount of Mg in proportion of Mg and Na would be expected. Applicant also alleges superior crystal quality however the is no comparative evidence of crystal quality.

Applicant's argument that Yoshida et al is directed to a multilayer process and Jeong et al is directed to a GaN monocrystal growth process is noted but not found persuasive. Jeong et al teaches Mg as an impurity in the melt. Yoshida et al is merely relied upon to teach a desirable amount of Mg in a GaN to produce desirable electrical properties. It is well known in the art of crystal growth that dopant is added to a melt to obtain a crystal with a desired amount of dopant (See Ivantzov et al col 6, In 1-25).

In response to applicant's argument that the examiner's conclusion of obviousness is based upon improper hindsight reasoning, it must be recognized that any judgment on obviousness is in a sense necessarily a reconstruction based upon hindsight reasoning. But so long as it takes into account only knowledge which was within the level of ordinary skill at the time the claimed invention was made, and does not include knowledge gleaned only from the applicant's disclosure, such a reconstruction is proper. See In re McLaughlin, 443 F.2d 1392, 170 USPQ 209 (CCPA 1971).

In response to applicant's argument that the addition of Mg improves reproducibility of the crystal growth, the fact that applicant has recognized another advantage which would flow naturally from following the suggestion of the prior art cannot be the basis for patentability when the differences would otherwise be obvious. See Ex parte Obiaya, 227 USPQ 58, 60 (Bd. Pat. App. & Inter. 1985). Applicant teaches on page 3 of the specification that the flux includes Mg and the crystal can have a p-type electrical characteristic and the amount of nitrogen dissolved in the melt is increased because the flux includes Mg (paragraph [0012]). Therefore, Applicant teaches the Mg has two functions, one to dope and one as a flux to increase amount of dissolved nitrogen. Ivantzov et all teaches doping with Mg to form a p-type GaN, where that amount of dopant overlaps the range taught by applicant. Therefore, the proportion of Mg to the sum of Mg and Na would naturally flow from the teachings of the prior art because the prior art teaches overlapping impurities concentrations in the grow crystal thus would require optimization of the amount of Mg and the amount of Mg is a dopant, thus would tend to be a small proportion of the melt mixture.